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Chapter 8 Chemical Equations Reactions Worksheet Answers

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Chapter 8 - Chemical Equations and Reactions

8 Chemical Equations and Reactions

Chapter 8 Chemical equations and reactions(no example

...

Chemistry Chapter 8- Chemical Equations and Reactions ...

Chapter 8 Chemical Equations

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Chapter 8 - Quantities
in Chemical Reactions

... Balancing Chemical
Equations Practice

Problems - Duration:

14:56. Tyler DeWitt

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14:56. Chapter 8 -
Basic Concepts of
Chemical ...
Reactions

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The usual evidence of an reaction is a solid precipitant, the evolution of a gas, or the formation of water. (Solubility rules).

Neutralization is the reaction b/t an acid (H^+ cation) and a base (OH^- anion) to form and

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ionic salt and water.

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Equations and
Reactions*

Chemical Equations
and Reactions The
evolution of energy as
light and heat is an
indication that a
chemical reaction is
taking place. CHAPTER
8 Fireworks

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Here we have given
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1 Chemical Reactions
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According to new CBSE
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Questions for Class 10

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*CHAPTER 8 Chemical
Equations and
Reactions*

Chapter 8 - Chemical
Equations and
Reactions. Describing
Chemical Reactions

_____ - process in which
one or more
substances are
changed into one or
more different
substances. Reactants
- original substance.
Products - resulting
substance . Described
by _____ Chemical

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Chemical Equations Reactions
equation - represents,
with symbols and
formulas, the identities
and ...

Worksheet

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Equations and
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8.1 The chemical
equation reactants -
the substances
entering the reaction
products - the
substances formed In a
chemical reaction
atoms are neither
created nor destroyed.

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All atoms present in the reactants must also be present in the products ex. thermite reaction

$$1. \text{Al} + \text{Fe}_2\text{O}_3 \rightarrow \text{Fe} + \text{Al}_2\text{O}_3$$

reactants

$$\text{products } 2. 2\text{Al} + \text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + \text{Al}_2\text{O}_3$$

CHAPTER 8 Chemical Equations and Reactions

chemical reaction in which one element replaces another element in a compound

double-displacement

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reaction a reaction in which a gas, a solid precipitate, or a molecular compound forms from the exchange of ions between two compounds

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A word equation and a formula equation are both qualitative.

Although they both show the reactants and

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products of a chemical reaction, they give no information about the amounts of reactants or products. A chemical equation, however, does show the relative amounts of the reactants and products in a chemical reaction.

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Chemical Equations Worksheet Answers

Equation must represent known facts.

2. The Equation must contain the correct formulas for the reactants and products.

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1. the coefficients of a chemical reaction indicate relative, not absolute, amounts of reactants and products

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2. the relative masses of the reactants and products of a chemical reaction can be determined from the reaction's coefficients
3. the reverse reaction for a chemical equation has the same relative amounts of substances as the forward ...

Chapter 8: Chemical Equations and Reactions Jeopardy Template

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Chemical Equations & Reactions.

We will conclude the chapter by learning about the different types of chemical reactions; students will be able to differentiate between synthesis, decomposition, single replacement, and double replacement reactions and be able to predict the products (if any) of a given reaction.

Chapter 8: Chemical
Page 16/25

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bc any instructions shown in a chemical equation can help you be sure the reaction turns out the way it should. when info about conditions of a reaction is desired, the _____ shows it!

arrow (ex: $N_2(g) + 3H_2(g) \xrightarrow[2500\text{ kPa}]{350^\circ\text{C}} 2NH_3(g)$)

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Chapter 8 - Chemical Equations and Reactions

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8 Chemical Equations and Reactions

formula equation for a given chemical

reaction. Balance a

formula equation by

inspection. FIGURE 8-1

The decomposition of ammonium dichromate proceeds rapidly, releasing energy in the form of light and heat.

Chapter 8 Chemical equations and reactions(no example

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Chapter 8: Chemical Equations and Reactions. Doing so actually changes the chemical formula, and that does not happen. Adjust the number of atoms on each side of the equation by changing the coefficients, the larger number in front of the compound. Check to make sure the number and type of each atom is the same on both sides of the equation.

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*Chemistry Chapter 8-
Chemical Equations
and Reactions ...*

CHAPTER 8 Chemical
Equations and
Reactions. SECTION 1
Describing Chemical
Reactions. OBJECTIVES.

1. List three observations that suggest that a chemical reaction has taken place.
2. List three requirements for a correctly written chemical equation.
- 3.

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Write a word equation for a correctly written chemical equation.

Chapter 8 Chemical Equations

CHAPTER 8 REVIEW

Chemical Equations and Reactions SECTION 2 SHORT ANSWER

Answer the following questions in the space provided. 1. Match the equation type on the left to its representation on the right. c synthesis (a)

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Chemical Equations Worksheet Answers

AX BY \rightarrow AY BX d
decomposition (b) A BX
 \rightarrow AX B b single-
displacement (c) A B \rightarrow
AX a double-
displacement (d) AX \rightarrow
A X

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