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Choose the best answer and write its letter on the line. 11. The fourth principal energy level has a. 4 orbitals. b. 16 orbitals. c. 32 orbitals. . 9 orbi s. 12. If the electron configuration of an element is  $1s^2 2s^2 2p^6 3s^2 3p^5$ , the 116 element is a. iron. b. bromine. Core Teaching Resources c. chlorine. d. phosphorus.

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(the answer is not an electron configuration, even though it is) ...  
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## **05 CTR ch03 7/9/04 3:25 PM Page 55 MEASUREMENTS AND THEIR ...**

Valence and core Electrons Name Date O O Purpose To discover the arrangements of electrons within atoms. Complete the table on the handout, filling in the missing atoms. Then answer the questions. I. How does the number of electrons change as you move from left to right a period? ... Living By Chemistry Teaching and Classroom Resources @ 2012 W ...

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