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Engineering Chemistry - May 2013. First Year Engineering (Set A) (Semester 1) TOTAL MARKS: 100 TOTAL TIME: 3 HOURS (1) Question 1 is compulsory. (2) Attempt any four from the remaining questions. (3) Assume data wherever required.

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ENGINEERING CHEMISTRY [BE-101] Time: 3Hrs Max Marks: 100 Min Marks: 35. Note: Attempt any 5 questions taking 1 question from each unit. All question Carry equal marks. Q.1.(a) Explain the principal and process of Lime- soda softening of water giving the different chemical reactions involved in the process.

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UNIT 1: Sources, Impurities, Hardness & its units, Determination of hardness by EDTA method, Alkalinity & its determination, Boiler troubles (Sludge & Scale, Priming & Foaming, Boiler Corrosion, Caustic Embrittlement), Softening of water by Lime-Soda, Zeolite and Ion Exchange methods, Internal treatment methods of Boilers, Numerical problems based on softening methods, hardness and alkalinity.

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Chemistry, Conductance-electrolyte in solution (Specific conductivity,Equivalent Conductivity and Molar Conductivity), Variation of specific conductivity with dilution, etc.

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