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Review Previous Concepts Gases CHAPTER 11 1. What is a mole? What information does the molar mass of a substance relate about the substance? 2. What is the kinetic-molecular theory of matter? Key Concepts LOUISIANA STANDARDS LA.PS.29 Predict the properties of a gas based on gas laws. (e.g., temperature, pressure, volume). (PS-H-C7) 340 CHAPTER 11

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the rates of effusion of gases at the same temperature and pressure are inversely proportional to the square roots of their molar masses Rate of effusion of A / B = $\sqrt{M_b} / \sqrt{M_a}$

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CHAPTER 11 REVIEW Molecular Composition of Gases MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The average speed of a gas molecule is most directly related to the . (a) polarity of the molecule (b) pressure of the gas (c) temperature of the gas (d) number of moles in the sample 2.

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You can test your readiness to proceed by answering the Review Questions at the end of the chapter. This might also be a good time to read the Chapter Objectives, which precede the Review Questions. ... 414 Chapter 11 Modern Atomic Theory. 11.1 The Mysterious Electron 415 When a guitar string is plucked, the string vibrates up and down in a ...

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