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Chemistry. All elements are composed of
atoms and they are indivisible. Atoms of
the same element are identical. atoms
different elements are different.

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The main purpose of the book is to help

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you learn chemistry. However, the book is also linked to two other goals: to show how important and interesting the subject is and to show how to think like a chemist. To solve complicated problems, the chemist uses logic, trial and error, intuition, and, above all, patience.

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The smallest particle of an element that retains its identity... The smallest particle of an element that retains its identity... A weighted average mass of the atoms in a naturally occurring... A

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stream of electrons produced at the negative electrode (cath... Atom The smallest particle of an element that retains its identity... Atomic mass unit (amu)...

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paper 1 Back A level paper 2 Back A
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Created by: jstoverinkQCHS TEACHER. 1. Describe the evolution of atomic theory leading to the current model of the atom based on the works of Democritus, Dalton, Thomson, Rutherford, Bohr, and Schrodinger. 2. Explain the law of ...

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Atomic Mass & Mass Number Atomic mass and mass number are not the same. Atomic mass is the atom's actual mass • 1 atomic mass unit (amu) is equal to 1/12 of the mass of the carbon-12 atom • a proton has a mass of 1.007 amu • a neutron has a mass of about 1.008 amu • an electron has a very small mass of 0.00055 amu

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simple knowledge of the subatomic particles to the more complex electron arrangements in orbitals. It is important to

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_____ atomic theory included the ideas:

- 1) Elements are composed of atoms,
- 2) Atoms of different elements are different,
- 3) Atoms of different elements can combine in simple whole-number ratios to form compounds,
- and 4)

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Chemical reactions occur when atoms are separated, joined, or rearranged.

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